

**Table 198:**  $b(E) \times 10^6$  [ $\text{cm}^2\text{g}^{-1}$ ] for  
Methanol ( $\text{CH}_3\text{OH}$ )  
 $\langle Z/A \rangle = 0.56176$

E [GeV]	$b_{\text{brems}}$	$b_{\text{pair}}$	$b_{\text{nucl}}$	$b_{\text{tot}}$
2.	0.2608	0.1134	0.4749	0.8491
5.	0.3540	0.2812	0.5024	1.1376
10.	0.4314	0.4272	0.4870	1.3456
20.	0.5132	0.5863	0.4642	1.5637
50.	0.6250	0.8100	0.4393	1.8744
100.	0.7088	0.9684	0.4271	2.1043
200.	0.7884	1.1146	0.4208	2.3238
500.	0.8829	1.2658	0.4198	2.5686
1000.	0.9438	1.3588	0.4265	2.7291
2000.	0.9945	1.4245	0.4380	2.8570
5000.	1.0454	1.4827	0.4594	2.9876
10000.	1.0728	1.5101	0.4811	3.0641
20000.	1.0921	1.5274	0.5068	3.1263
50000.	1.1090	1.5411	0.5468	3.1968
100000.	1.1168	1.5466	0.5813	3.2448